

Buffer pKa and pH Range Values

Buffers	pKa	For preparation of Buffers in the pH range
Hydrochloric Acid - HCl		0-2
Nitric Acid - HNO ₃		
Perchloric Acid – HClO ₄		
Potassium Chloride – KCl		1.1-1.8
Oxalic Acid – C ₂ H ₂ O ₄	1.3, 4.3	0.3-5.3
Sodium Oxalate – C ₂ Na ₂ O ₄		
Potassium Tetraoxalate		
Maleic Acid – C ₄ H ₄ O ₄	1.9, 6.2	0.9-2.9, 5.2-7.2
Phosphoric Acid – H ₃ PO ₄	2.1, 7.2, 12.4	1.1-3.1, 6.2-10.1
Potassium Phosphate, Monobasic – H ₂ KO ₄ P		
Potassium Phosphate, Dibasic – HK ₂ PO ₄		
Potassium Phosphate, Tribasic – K ₃ PO ₄		
Sodium Phosphate, Monobasic – H ₂ NaPO ₄		
Sodium Phosphate, Dibasic – HNa ₂ PO ₄		
Sodium Phosphate, Tribasic – Na ₃ PO ₄		
Glycine – C ₂ H ₅ NO ₂	2.4, 9.7	1.4-3.4, 8.7-10.7
Monochloroacetic Acid – C ₂ H ₃ ClO ₂	2.8	1.8-3.8
Chloroacetic Acid, Sodium Salt ↑		
Phthalic Acid – C ₈ H ₆ O ₄	2.9, 5.4	1.9-6.4
Potassium Bipthalate		
d-Tartaric Acid – C ₄ H ₆ O ₆	3.0, 4.4	2.0-5.4
Potassium Tartrate – C ₄ H ₄ K ₂ O ₆		
Citric Acid – C ₆ H ₈ O ₇	3.1, 4.8, 6.4	2.1-7.4
Potassium Citrate – C ₆ H ₅ K ₃ O ₇		
Formic Acid – CH ₂ O ₂	3.8	2.8-4.8
Sodium Formate – CHNaO ₂		

Buffer pKa and pH Range Values Continued

Buffers	pKa	For preparation of Buffers in the pH Range
Succinic Acid – C ₄ H ₆ O ₄	4.2, 5.6	3.2-6.6
Acetic Acid – C ₂ H ₄ O ₂	4.6	3.6-5.6
Sodium Acetate – C ₂ H ₃ NaO ₂		
Ammonium Acetate – C ₂ H ₇ NO ₂	4.6(Acetic Acid) 9.3(Ammonium Hydroxide)	6-8
Hexamethylenetetramine – C ₆ H ₁₂ N ₄	5.1	4.1-6.1
Pyridine – C ₅ H ₅ N	5.2	4.2-6.2
(Carbonic Acid) – CH ₂ O ₃	6.3, 10.3	5.3-7.3, 9.3-11.3
Sodium Bicarbonate – CHNaO ₃		
Sodium Carbonate – CNa ₂ O ₃		
Potassium Bicarbonate – CKO ₃		
Potassium Carbonate – CK ₂ O ₃		
Ethylenediamine – C ₂ H ₈ N ₂	7.4, 10.2	6.4-8.4, 9.2-11.2
Imidazole – C ₃ H ₃ N ₂	7.5	6.5-8.5
Triethanolamine – C ₆ H ₁₅ NO ₃	7.8	6.8-8.8
TRIS Hydrochloride – C ₄ H ₁₂ ClNO ₃		
HEPES (4-(2-Hydroxyethyl)-1-piperazineethanesulfonic Acid) 0 C ₈ H ₁₈ N ₂ SO ₄		
2-Amino-2-methyl-1,2-propanediol. (AMP) – C ₄ H ₁₁ NO	8.8	7.8-9.8
Boric Acid – H ₃ BO ₃	9.2	8.2-10.2
Sodium Borate – B ₄ Na ₂ O ₇		
Ammonium Hydroxide – H ₅ NO	9.3	8.3-10.3
Ammonium Chloride – ClH ₄ N		
Diethylamine – C ₄ H ₁₁ N	11.0	10.0-12.0
Calcium Hydroxide – CaO ₂ H ₂		12.4 (sat. soln)
Potassium Hydroxide, 45% soln - KHO		12-14
Sodium Hydroxide, 50% soln - NaOH		